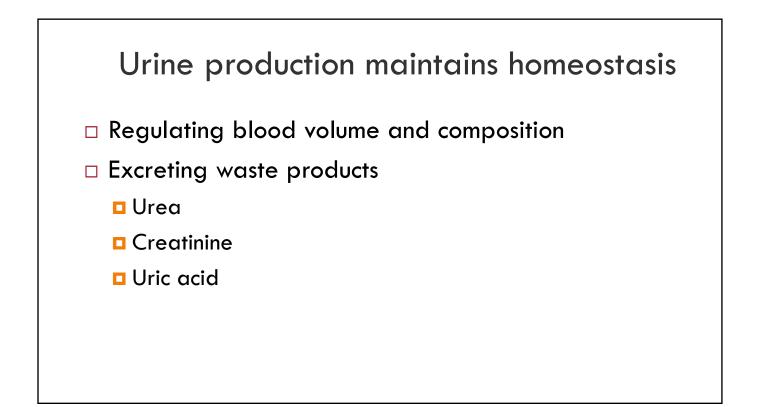
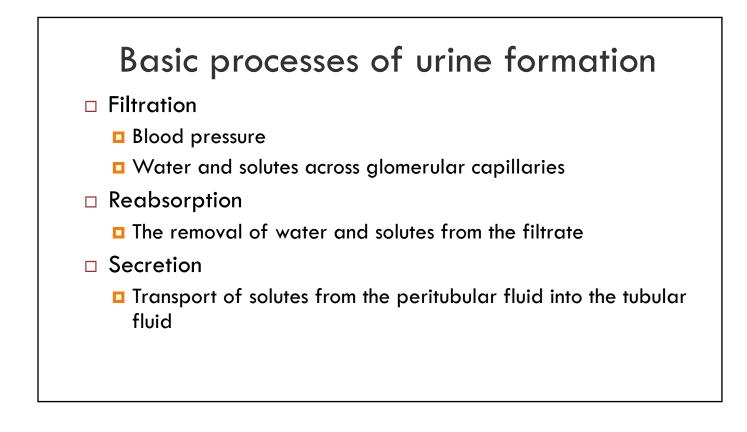
RENAL PHYSIOLOGY

Danil Hammoudi.MD

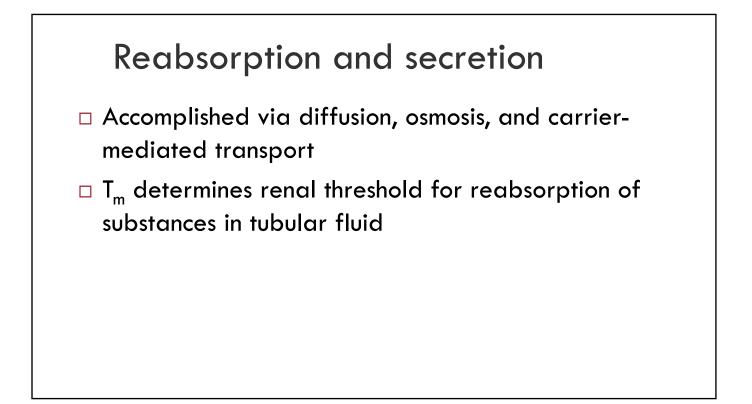
Functions

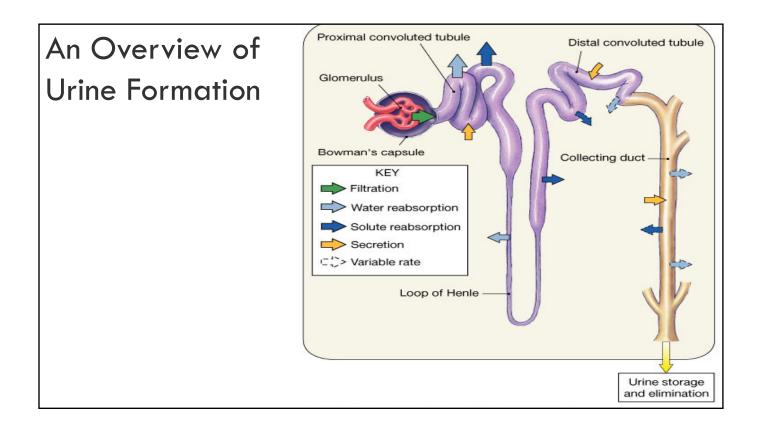
- Regulating blood ionic composition
- Regulating blood pH
- Regulating blood volume
- Regulating blood pressure
- Produce calcitrol and erythropoietin
- Regulating blood glucose
- Excreting wastes

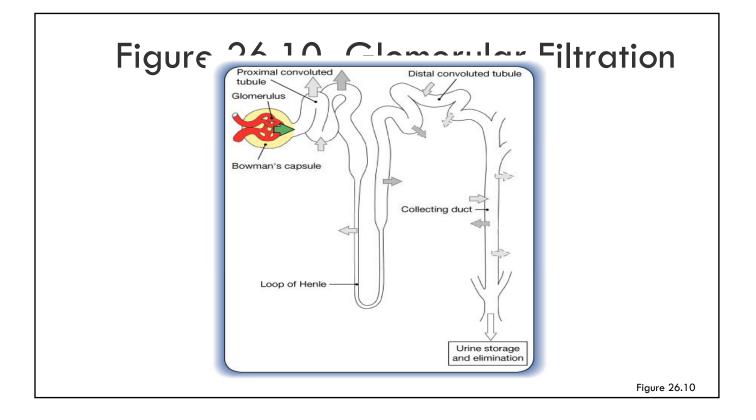


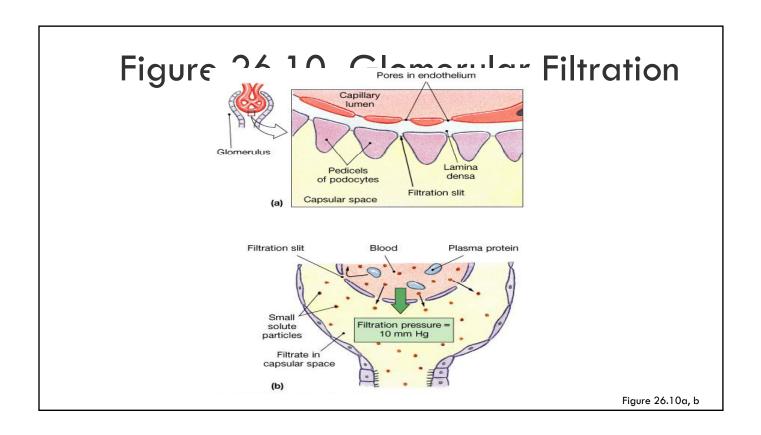


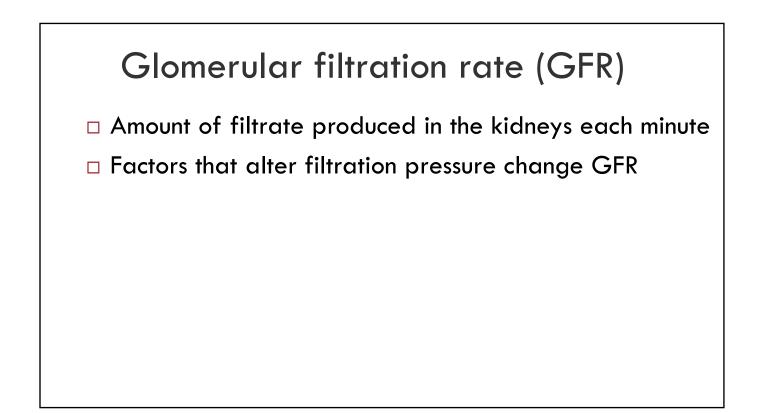
Carrier Mediated Transport Filtration in the kidneys modified by carrier mediated transport Facilitated diffusion Active transport Cotransport Countertransport Carrier proteins have a transport maximum (T_m) Determines renal threshold





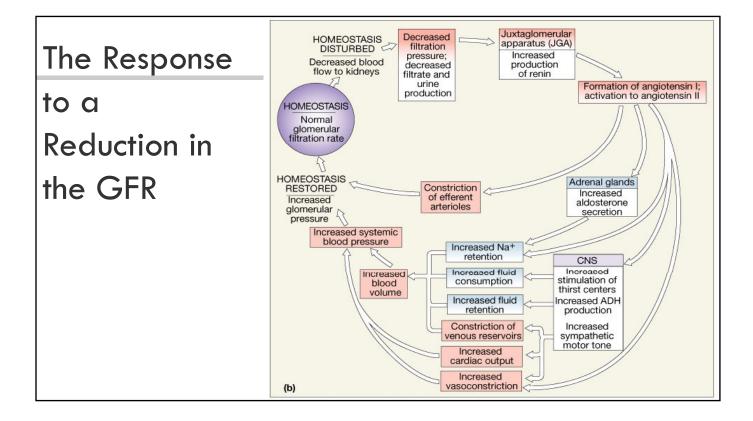


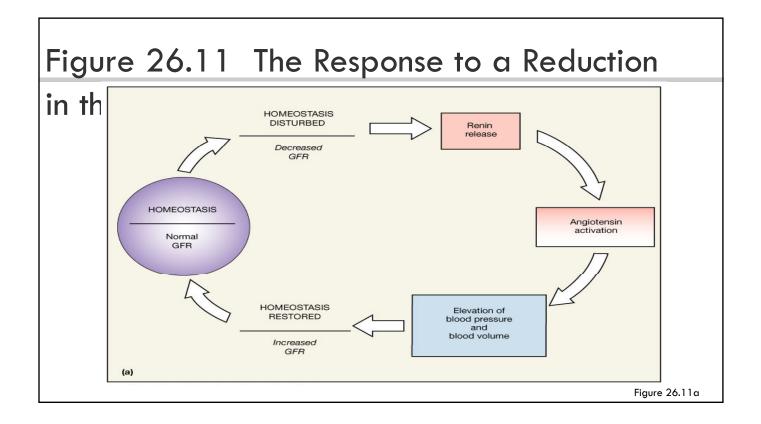


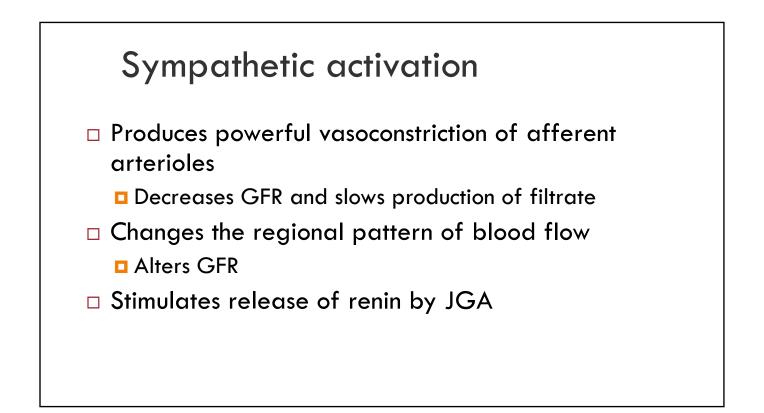


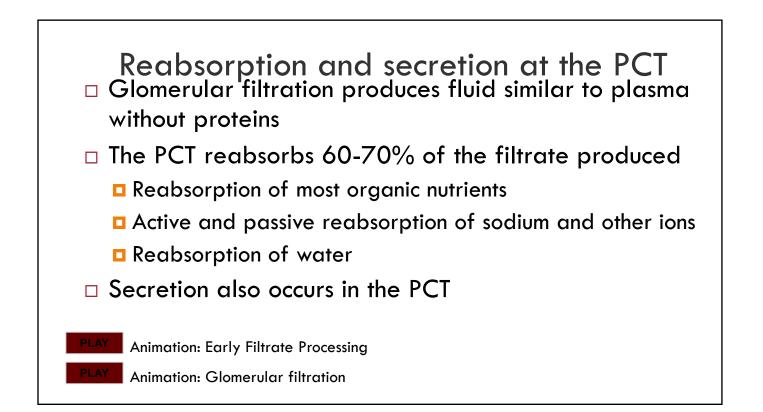
Factors controlling the GFR

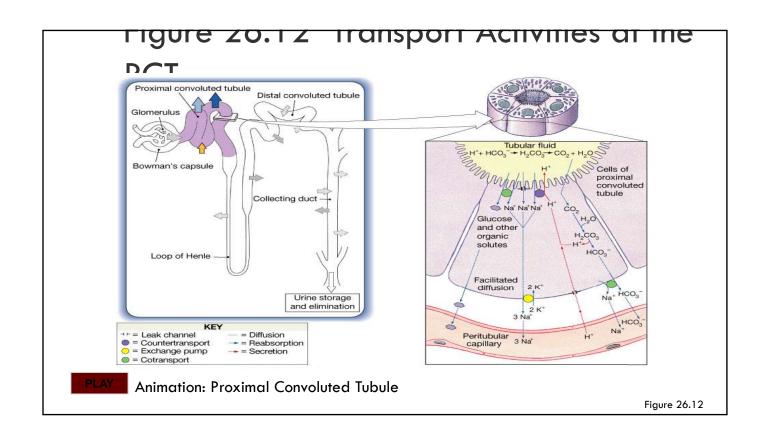
- A drop in filtration pressure stimulates Juxtaglomerular apparatus (JGA)
 - Releases renin and erythropoietin

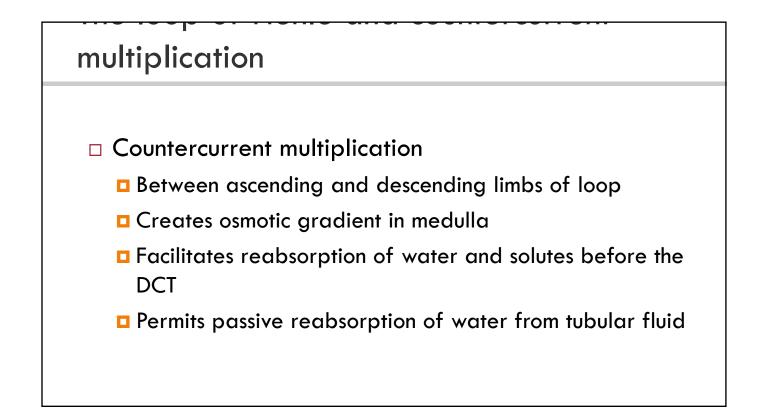


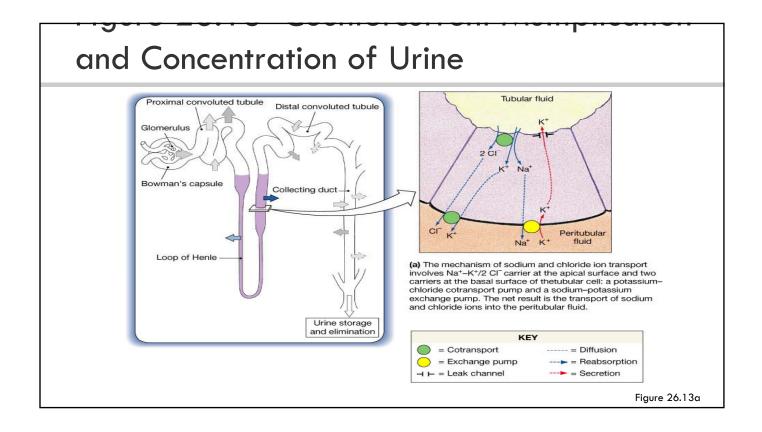


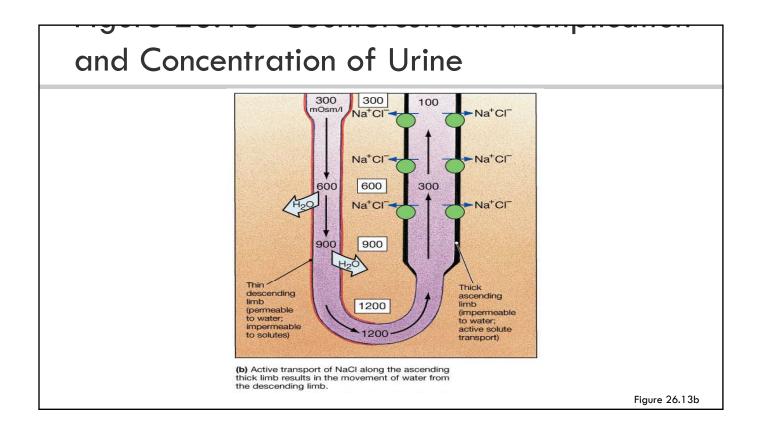


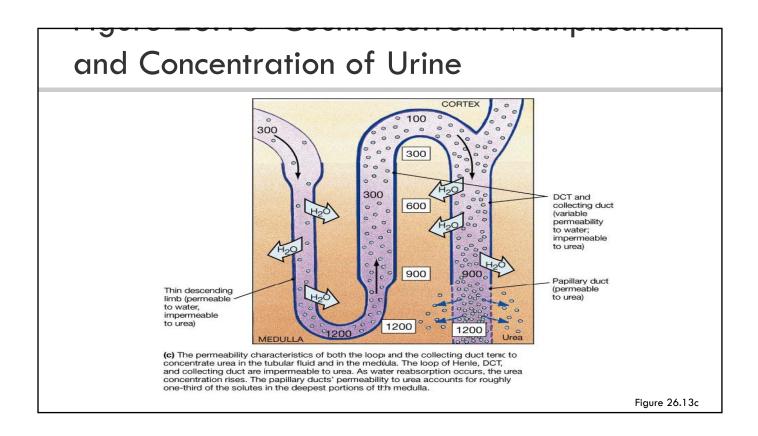


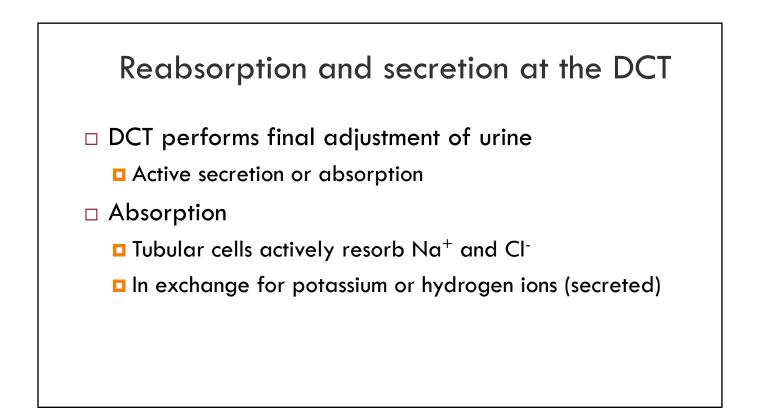


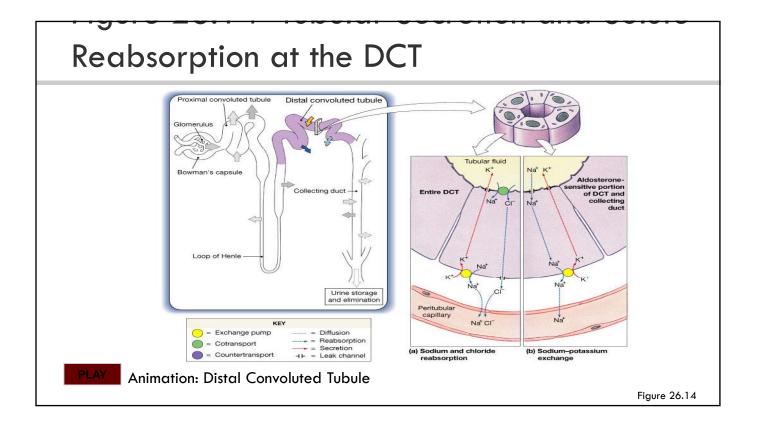


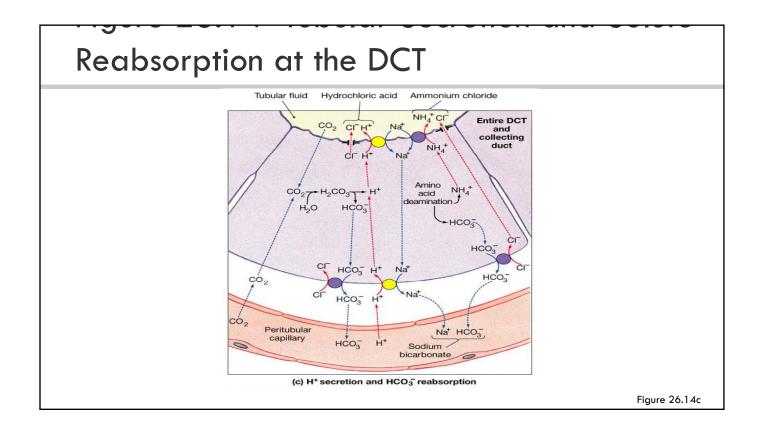


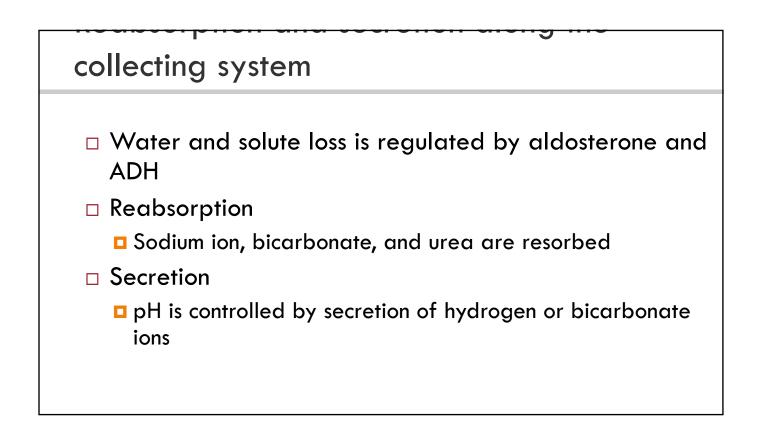






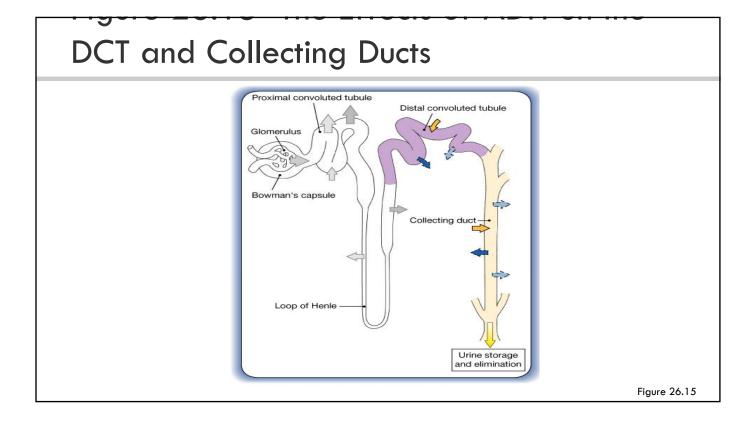


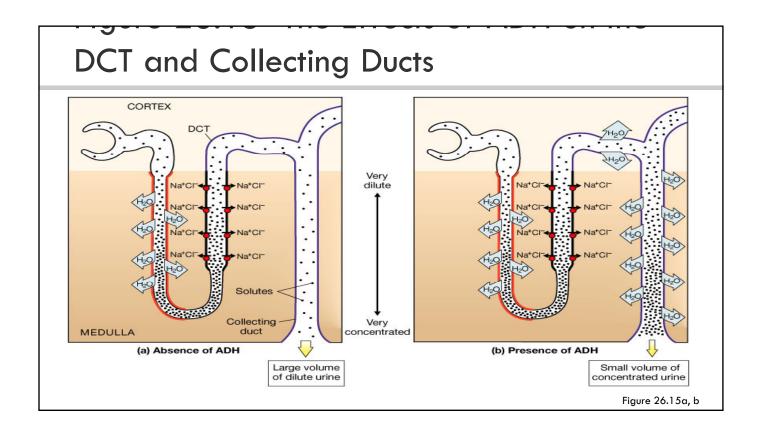


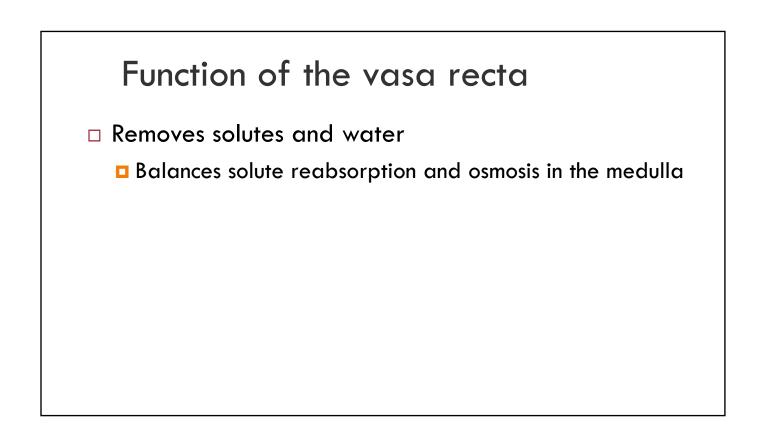


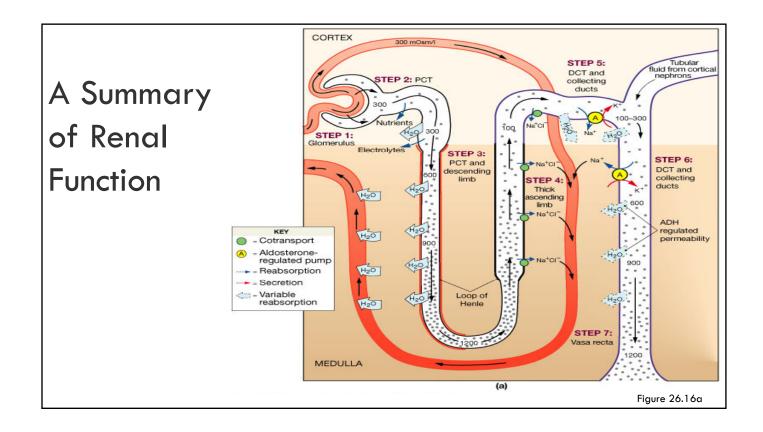
concentration

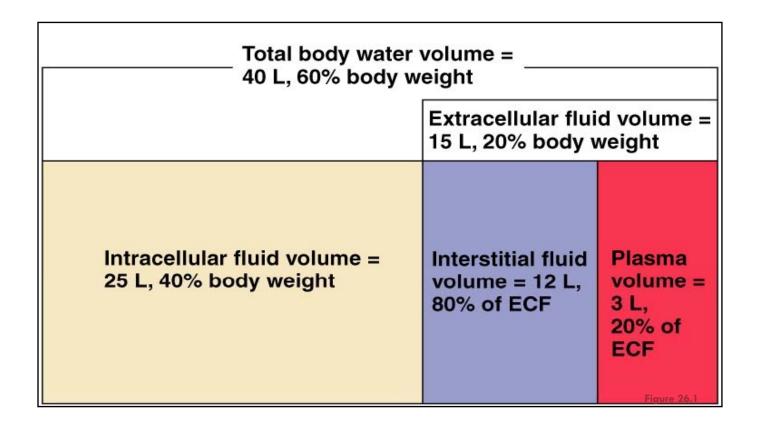
- Urine volume and osmotic concentration are regulated by controlling water reabsorption
 - Precise control allowed via facultative water reabsorption

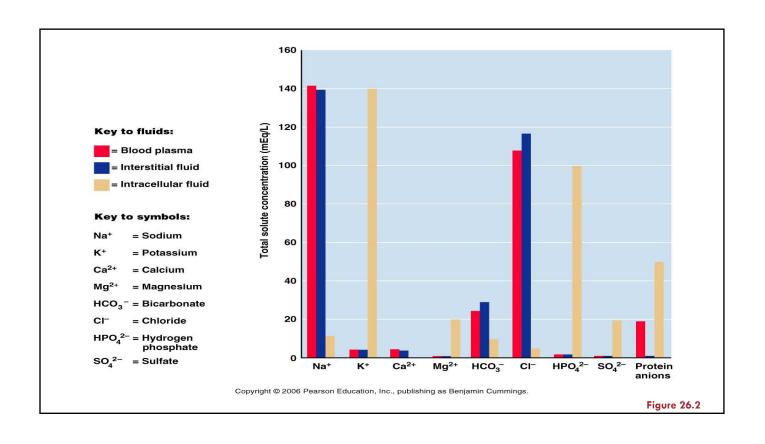


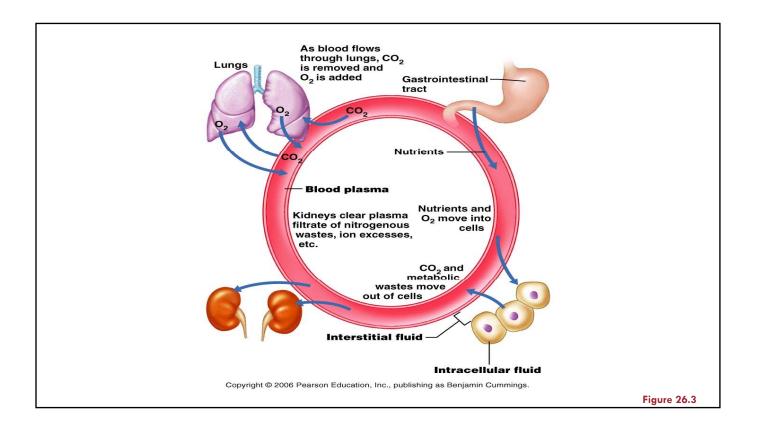


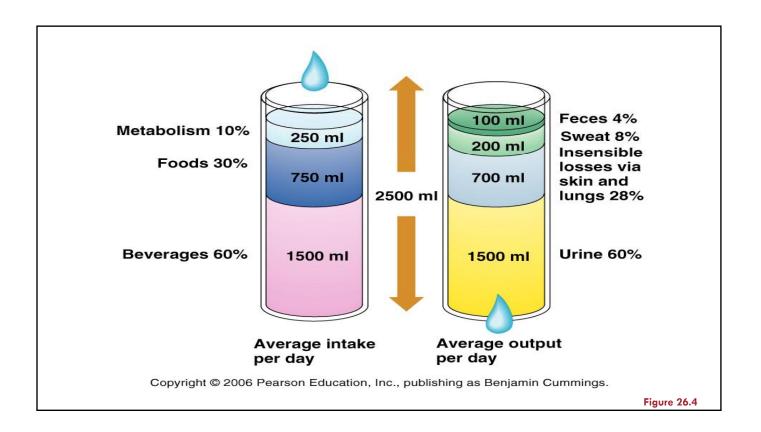


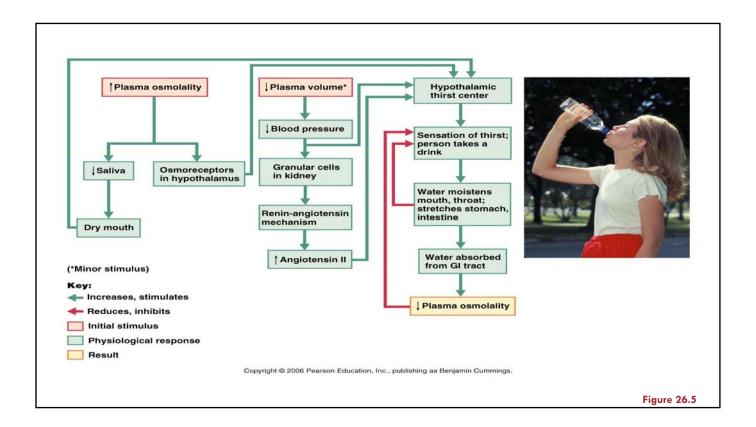


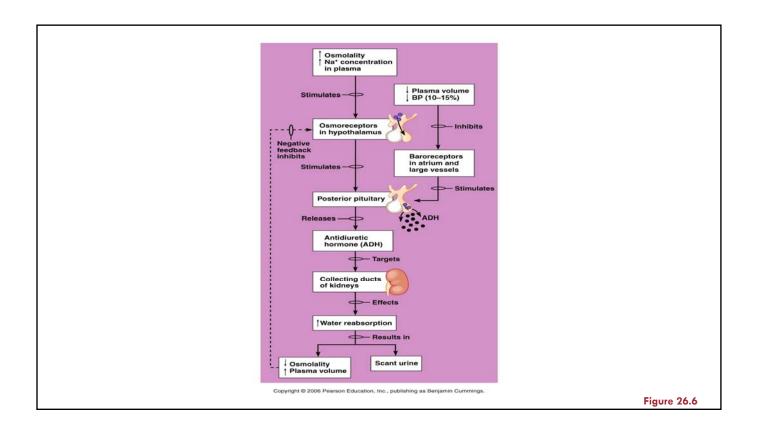


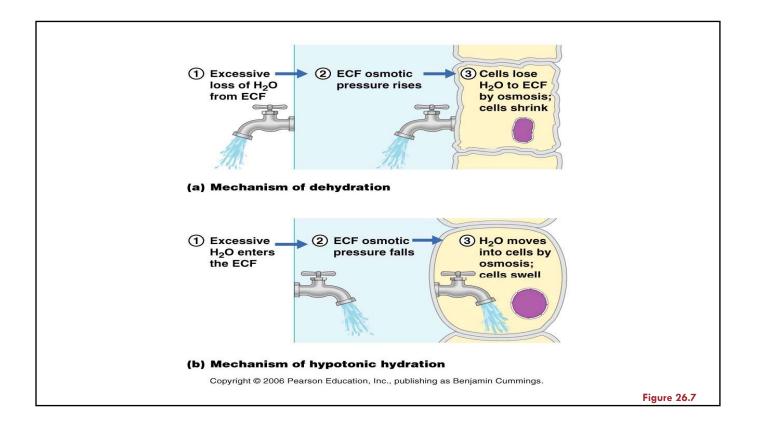


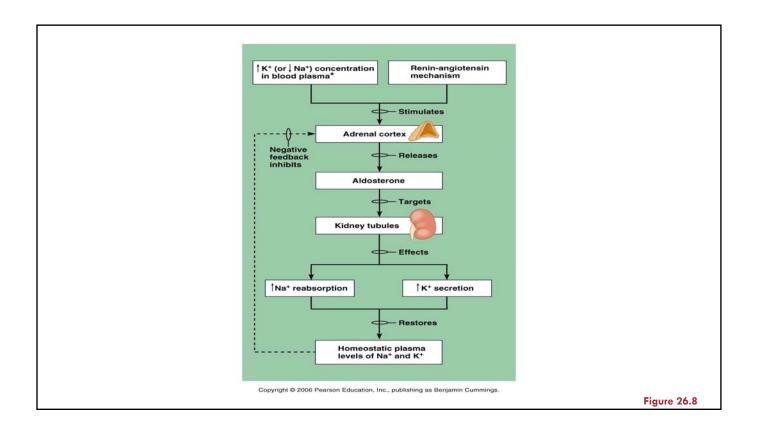


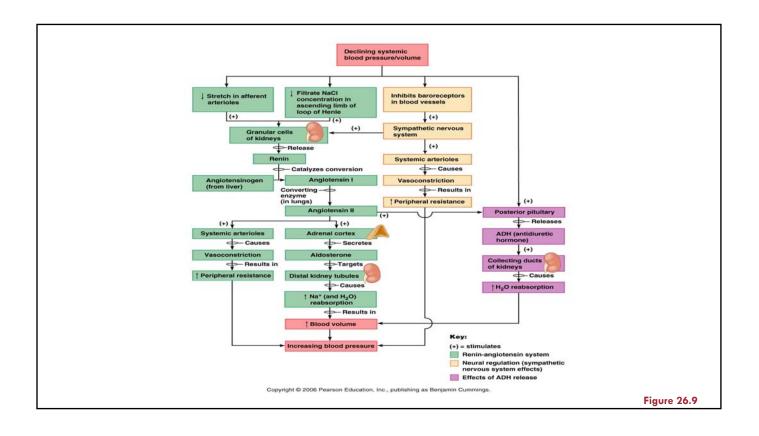


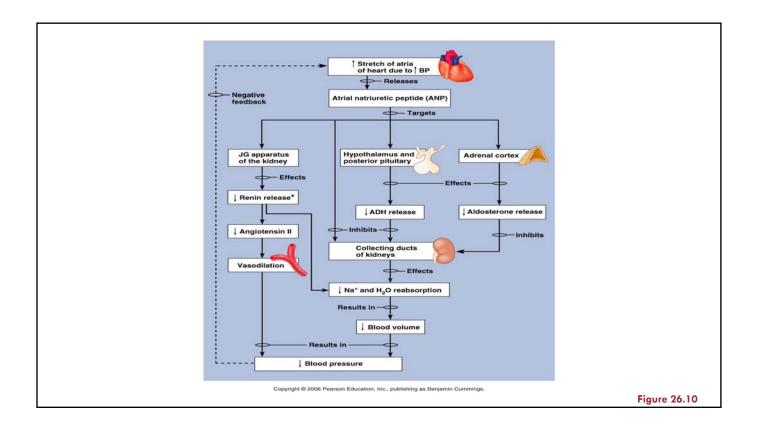


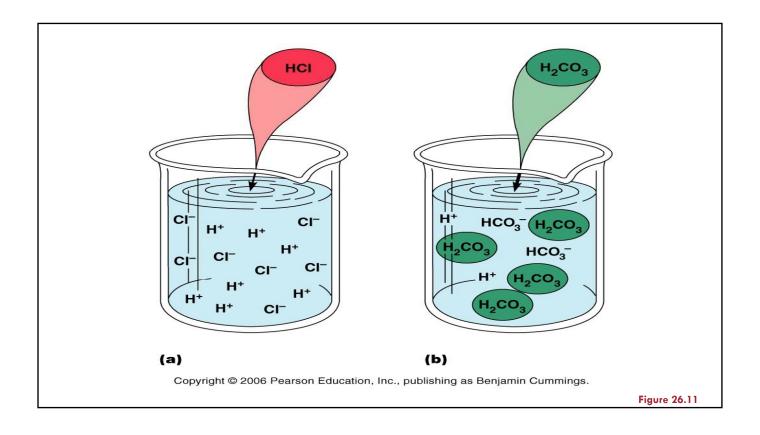


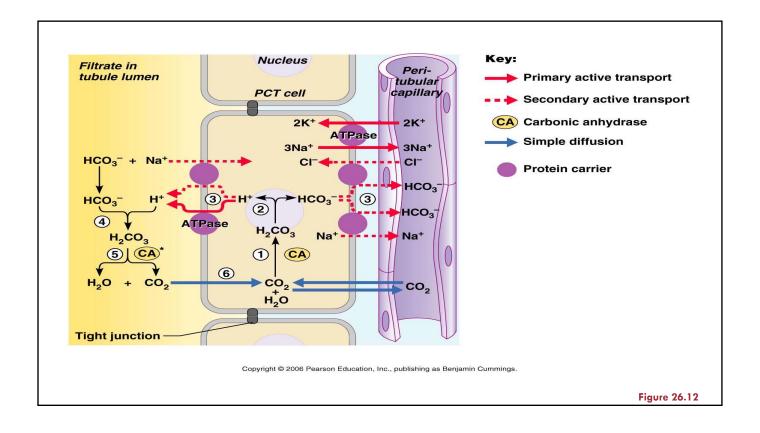


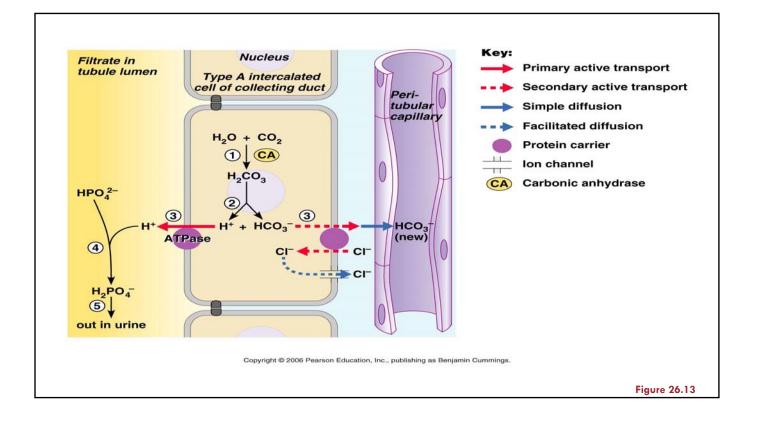


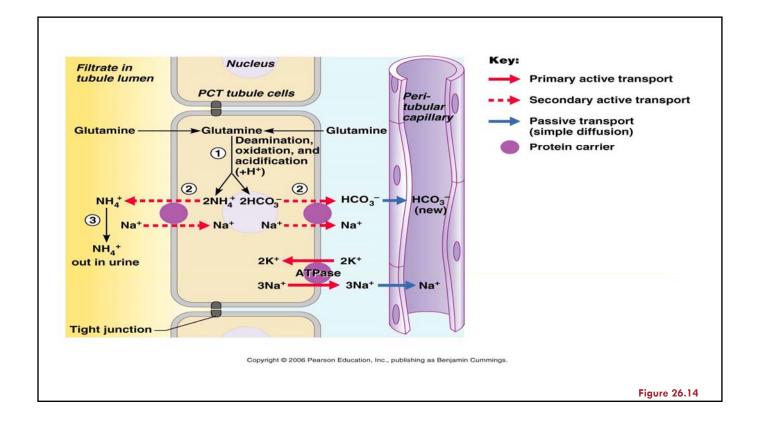










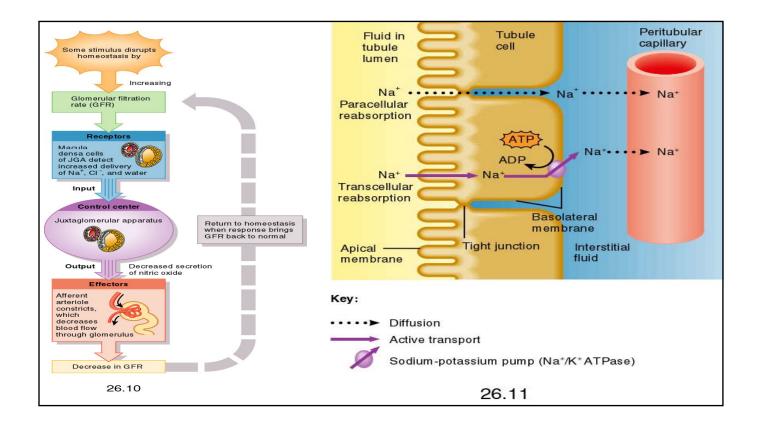


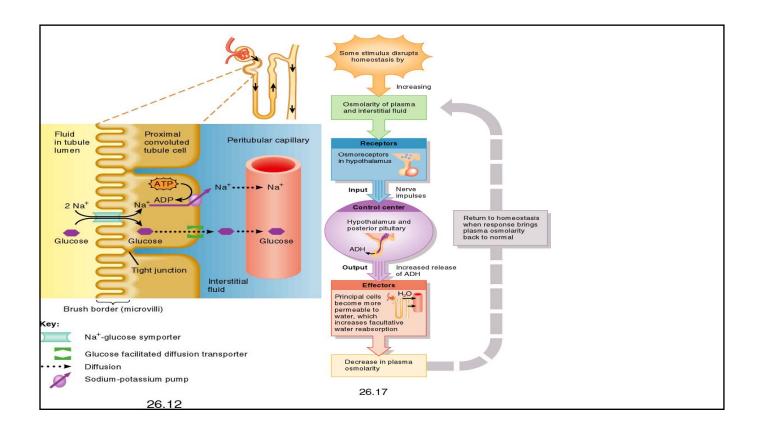
ION	ABNORMALITY (SERUM VALUE)	POSSIBLE CAUSES	CONSEQUENCES	
Sodium	Hypernatremia (Na ⁺ excess: >145 mEq/L)	Dehydration; uncommon in healthy individuals; may occur in infants or the confused aged (individuals unable to indicate thirst) or may be a result of excessive intravenous NaCl administration	Thirst: CNS dehydration leads to confusion and lethargy progressing to coma; increased neuromuscular irritability evidenced by twitching and convulsions	
	Hyponatremia (Na⁺ deficit: <135 mEq/L)	Solute loss, water retention, or both (e.g., excessive Na ⁺ loss through vomiting, diarrhea, burned skin, tubal drainage of stomach, and as a result of excessive use of diuretics); deficiency of aldosterone (Addison's disease); renal disease; excess ADH release; excess H ₂ O ingestion	Most common signs are those of neurologic dysfunction due to brain swelling. If sodium amounts are actually normal but water is excessive, the symptoms are the same as those of water excess: mental confusion; giddiness; coma if development occurs slowly; muscular twitching, irritability, and convulsions if the condition develops rapidly. In hyponatremia accompanied by water loss, the main signs are decreased blood volume and blood pressure (circulatory shock)	
Potassium	Hyperkalemia (K ⁺ excess: >5.5 mEq/L)	Renal failure; deficit of aldosterone; rapid intravenous infusion of KCI; burns or severe tissue injuries which cause K ⁺ to leave cells	Nausea, vomiting, diarrhea; bradycardia; cardiac arrhythmias, depression, and arrest; skeletal muscle weakness; flaccid paralysis	
	Hypokalemia (K ⁺ deficit: <3.5 mEq/L)	Gastrointestinal tract disturbances (vomiting, diarrhea), gastrointestinal suction; Cushing's disease; inadequate dietary intake (starvation); hyperaldosteronism; diuretic therapy	Cardiac arrhythmias, flattened T wave; muscular weakness; metabolic alkalosis; mental confusion; nausea; vomiting	
Phosphate	Hyperphosphatemia (HPO ₄ ^{2–} excess: >2.9 mEq/L)	Decreased urinary loss due to renal failure; hypoparathyroidism; major tissue trauma; increased intestinal absorption	Clinical symptoms arise because of reciprocal changes in Ca ²⁺ levels rather that directly from changes in plasma phosphate concentrations	
	Hypophosphatemia (HPO4 ²⁻ deficit: <1.6 mEq/L)	Decreased intestinal absorption; increased urinary output; hyperparathyroidism		

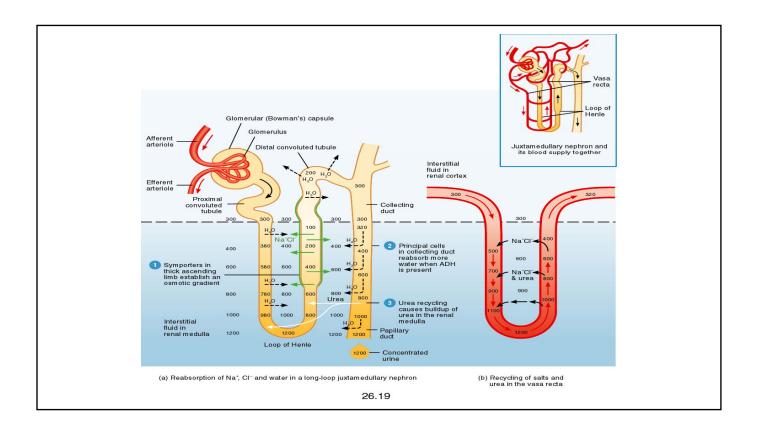
ION	ABNORMALITY (SERUM VALUE)	POSSIBLE CAUSES	CONSEQUENCES
Chloride	Hyperchloremia (Cl ⁻ excess: >105 mEq/L)	Dehydration; increased retention or intake; metabolic acidosis; hyperparathyroidism	No direct clinical symptoms; symptoms generally associated with the underlying cause, which is often related to pH abnormalities
	Hypochloremia (Cl ⁻ deficit: <95 mEq/L)	Metabolic alkalosis (e.g., due to vomiting or excessive ingestion of alkaline substances); aldosterone deficiency	
Calcium	Hypercalcemia (Ca ²⁺ excess: >5.2 mEq/L or 10.5 mg%)*	Hyperparathyroidism; excessive vitamin D; prolonged immobilization; renal disease (decreased excretion); malignancy	Decreased neuromuscular excitability leading to cardiac arrhythmias and arrest, skeletal muscle weakness, confusion, stupor and coma; kidney stones; nausea and vomiting
	Hypocalcemia (Ca ²⁺ deficit: <4.5 mEq/L or 9 mg%)*	Burns (calcium trapped in damaged tissues); hypoparathyroidism; vitamin D deficiency; renal tubular disease; renal failure; hyperphosphatemia; diarrhea; alkalosis	Increased neuromuscular excitability leading to tingling of fingers, tremors, skeletal muscle cramps, tetany, convulsions; depressed excitability of the heart; osteomalacia; fractures
Magnesium	Hypermagnesemia (Mg ²⁺ excess: >2.2 mEq/L)	Rare; occurs in renal failure when Mg is not excreted normally; excessive ingestion of Mg ²⁺ -containing antacids	Lethargy; impaired CNS functioning, coma, respiratory depression; cardiac arrest
	Hypomagnesemia (Mg ²⁺ deficit: <1.4 mEq/L)	Alcoholism; loss of intestinal contents, severe malnutrition; diuretic therapy	Tremors, increased neuromuscular excitability, tetany, convulsions

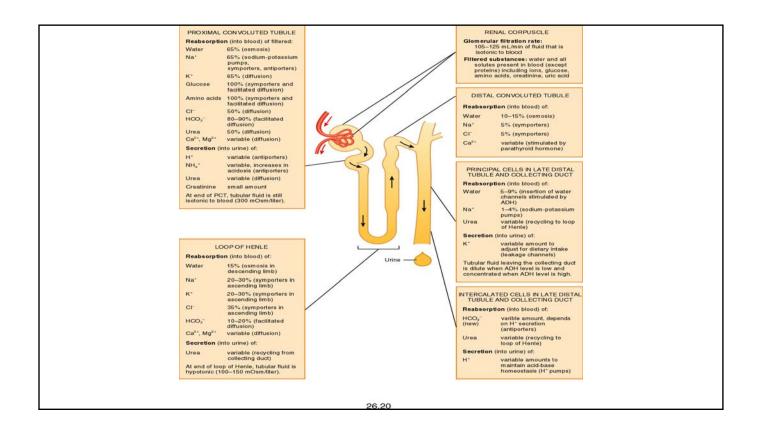
CONDITION AND HALLMARK	POSSIBLE CAUSES; COMMENTS		
METABOLIC ACIDOSIS			
uncompensated (uncorrected) (HCO3 ⁻ <22 mEq/L;	Severe diarrhea: bicarbonate-rich intestinal (and pancreatic) secretions rushed through digestive tract before their solutes can be reabsorbed; bicarbonate ions are replaced by renal mechanisms that generate new bicarbonate ions		
pH <7.35)	Renal disease: failure of kidneys to rid body of acids formed by normal metabolic processes		
	Untreated diabetes mellitus: lack of insulin or inability of tissue cells to respond to insulin, resulting in inability to use glucose; fats are used as primary energy fuel, and ketoacidosis occurs		
	Starvation: lack of dietary nutrients for cellular fuels; body proteins and fat reserves are used for energy—both yield acidic metabolites as they are broken down for energy		
	Excess alcohol ingestion: results in excess acids in blood		
	High ECF potassium concentrations: potassium ions compete with H ⁺ for secretion in renal tubules; when ECF levels of K ⁺ are high, H ⁺ secretion is inhibited		
METABOLIC ALKALOSIS			
uncompensated (HCO ₃ >26 mEq/L;	Vomiting or gastric suctioning: loss of stomach HCl requires that H ⁺ be withdrawn from b to replace stomach acid; thus H ⁺ decreases and HCO ₃ increases proportionately		
pH >7.45)	Selected diuretics: cause K ⁺ depletion and H ₂ O loss. Low K ⁺ directly stimulates the tubule cells to secrete H ⁺ . Reduced blood volume elicits the renin-angiotensin mechanism, which stimulates Na ⁺ reabsorption and H ⁺ secretion.		
	Ingestion of excessive sodium bicarbonate (antacid): bicarbonate moves easily into ECF, where it enhances natural alkaline reserve		
	Excess aldosterone (e.g., adrenal tumors): promotes excessive reabsorption of Na ⁺ , which pulls increased amount of H ⁺ into urine. Hypovolemia promotes the same relative effect because aldosterone secretion is increased to enhance Na ⁺ (and H ₂ O) reabsorption.		

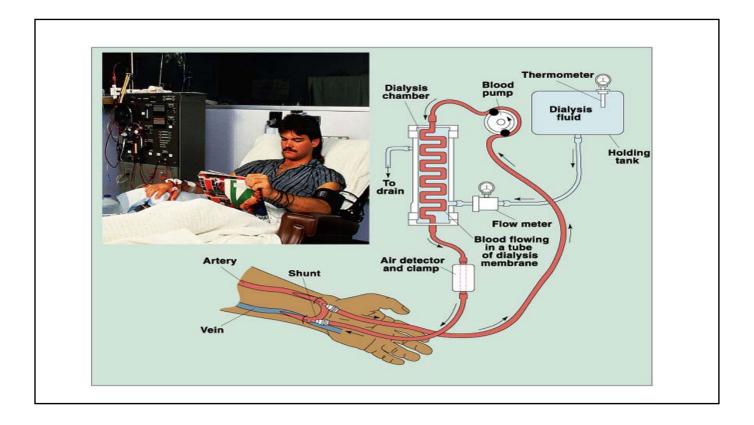
CONDITION AND HALLMARK	POSSIBLE CAUSES; COMMENTS		
RESPIRATORY ACIDOSIS	(HYPOVENTILATION)		
uncompensated (>45 mm Hg;	Impaired lung function (e.g., in chronic bronchitis, cystic fibrosis, emphysema): impaired gas exchange or alveolar PCO ₂ ventilation		
pH <7.35)	Impaired ventilatory movement: paralysis of respiratory muscles, chest injury, extreme obesity		
	Narcotic or barbiturate overdose or injury to brain stem: depression of respiratory centers, resulting in hypoventilation and respiratory arrest		
RESPIRATORY ALKALOS	S (HYPERVENTILATION)		
uncompensated	Strong emotions: pain, anxiety, fear, panic attack		
(P _{CO₂} <35 mm Hg; pH >7.45)	Hypoxia: asthma, pneumonia, high altitude; represents effort to raise P_{O_2} at the expense of excessive CO_2 excretion		
	Brain tumor or injury: abnormality of respiratory controls		
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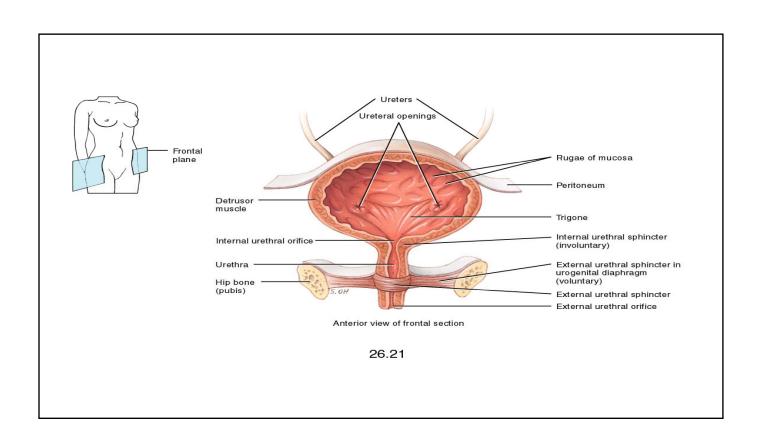


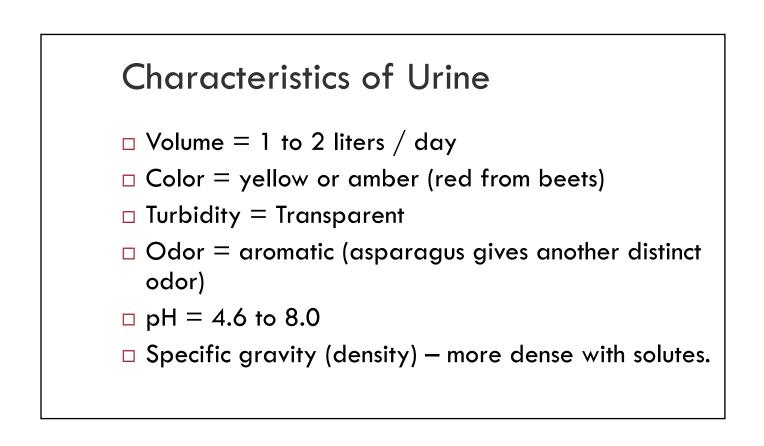












	or Reabsorb	
<u>Substance</u>	Plasma	Urine
Proteins	200 g	0.1 g
Bicarbonate	4.6 g	0
Glucose	3 g	0
Urea	4.8 g	25 g
Uric acid	0.15 g	0.8 g
Creatinine	0.03 g	1.6 g

Normal Constituents of Urine

- Urea from metabolism of amino acids
- □ Creatinine from creatine metabolism
- □ Uric acid from catabolism of nucleic acids
- Urobilinogen breakdown of hemoglobin
- Hippuric acid, indican, and ketone bodies
- Other substances and inorganic molecules

Abnormal Constituents of Urine

- Albuminuria
- Glucosuria
- Hematuria
- Pyuria
- Ketonuria (Ketosis)
- Bilirubinuria
- Casts
- Renal calculi
- Microbes

Words

- □ Glomerulonephritis (Steptococcus)
- Pyelonephritis
- □ Cystitis
- Polycystic disease
- Renal failure
- □ UTI's
- Azotemia

More Words

Intravenous Pyelogram (IVP)

- 🗆 Uremia
- Urethritis
- Enuresis
- Nocturnal enuresis
- □ Micturition
- □ Incontinence

More Words

- Lithotripsy
- Dysuria
- Polyuria
- Nocturia
- Anuria
- Oliguria

